

## Research paper

# Merging memantine and ferulic acid to probe connections between NMDA receptors, oxidative stress and amyloid- $\beta$ peptide in Alzheimer's disease

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## ABSTRACT

N-methyl-D-aspartate receptors (NMDAR) are critically involved in the pathogenesis of Alzheimer's disease (AD). Acting as an open-channel blocker, the anti-AD drug memantine preferentially targets NMDAR overactivation, which has been proposed to trigger neurotoxic events mediated by amyloid  $\beta$  peptide (A $\beta$ ) and oxidative stress. In this study, we applied a multifunctional approach by conjugating memantine to ferulic acid, which is known to protect the brain from A $\beta$  neurotoxicity and neuronal death caused by ROS. The most interesting compound (**7**) behaved, like memantine, as a voltage-dependent antagonist of NMDAR (IC<sub>50</sub> = 6.9  $\mu$ M). In addition, at 10  $\mu$ M concentration, **7** exerted antioxidant properties both directly and indirectly through the activation of the Nrf-2 pathway in SH-SY5Y cells. At the same concentration, differently from the parent compounds memantine and ferulic acid alone, it was able to modulate A $\beta$  production, as revealed by the observed increase of the non-amyloidogenic sAPP $\alpha$  in H4-SW cells. These findings suggest that compound **7** may represent a promising tool for investigating NMDAR-mediated neurotoxic events involving A $\beta$  burden and oxidative damage.

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## 1. Introduction

Synaptic loss is a major feature in Alzheimer's disease (AD). This

chronic neurodegenerative condition, which is currently afflicting about 47 million people worldwide, slowly destroys neurons leading to progressive cognitive disabilities [1]. How synapses are affected in the disease process remains unclear. The mechanistic understanding of synaptic damage represents a challenging goal, and may offer new possibilities for the prevention and cure of the disease. N-methyl-D-aspartate receptors (NMDAR) are ionotropic glutamate receptors known to play an important role for synaptic plasticity in the healthy brain [2]. They are primarily involved in neuronal excitatory synaptic transmission that underlies learning and memory but also in excitotoxic damage occurring during acute brain injuries and chronic neurodegenerative conditions. Targeting NMDAR therapeutically is therefore complicated by the dichotomous nature of their downstream signaling. It is the common view that these opposite effects depend on receptor localization: activation of synaptic NMDAR (sNMDAR) may contribute to cell

*Abbreviations:* AD, Alzheimer's disease; ANOVA, analysis of variance; APP, amyloid precursor protein; A $\beta$ , amyloid  $\beta$  peptide; DCFH-DA, dichlorodihydrofluorescein diacetate; eNMDAR, extrasynaptic NMDAR; FA, ferulic acid; fl-APP, full-length APP; H4-SW, APP-Swedish mutant-expressing H4 cell lines; HO-1, heme oxygenase-1; MEM, memantine; MTT, 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-tetrazolium bromide; NMDAR, N-methyl-D-aspartate receptors; Nrf2, nuclear factor (erythroid-derived 2)-like 2; Ppm, parts per millions; sAPP $\alpha$ , soluble amyloid precursor protein  $\alpha$ ; SDS, sodium dodecyl sulfate; sNMDAR, synaptic NMDAR; TMS, tetramethylsilane.

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survival and plasticity, while activation of extrasynaptic NMDAR (eNMDAR) may preferentially signal to neuronal death [3,4]. In particular, overactivation of eNMDAR has been associated with glutamate-mediated oxidative damage potentially leading to aberrant, misfolded proteins [5]. The amyloid  $\beta$  peptide ( $A\beta$ ) is a pathogenic feature of AD development. Produced by the sequential cleavage of the amyloid precursor protein (APP) by  $\beta$ - and  $\gamma$ -secretases, as an alternative to the non-amyloidogenic cleavage performed by  $\alpha$ -secretase,  $A\beta$  monomers aggregate into soluble oligomeric forms, which are believed to be mainly responsible for amyloid-driven synaptotoxicity [6]. A toxic positive feedback is established between  $A\beta$  production and eNMDAR overactivation, which involves cytoplasmic  $Ca^{2+}$  upregulation and aberrant redox-mediated reactions [7].

Memantine is an anti-AD drug currently in use for the treatment of moderate-to-severe forms of the disease. It is an uncompetitive/fast off-rate NMDAR antagonist. By acting as an open-channel blocker, it preferentially enters the channel's pore in conditions of excessive and prolonged glutamate exposure [8,9]. Its favorable kinetics has been proposed to selectively direct memantine's efficacy toward extrasynaptic/tonically-activated NMDAR over synaptic/phasicly-activated NMDAR [10], accounting for the clinical tolerability of the drug. Further, this peculiar profile seems to play a crucial role in determining memantine's ability to alleviate  $A\beta$ -induced synaptic dysfunction and to rescue both neuronal oxidative stress and the transient memory impairment caused by  $A\beta$  oligomers [11]. Unfortunately, however, like other available anti-AD drugs, memantine offers only a symptomatic relief to patients and is not able to halt the disease progression.

Based on these premises, we sought to combine in a single molecule memantine, which specifically modulates NMDAR-mediated excitotoxicity, responsible for ROS- and  $A\beta$ -mediated neurotoxic events, with the antioxidant ferulic acid (FA), whose well-established biological properties include the ability to protect the brain from  $A\beta$  neurotoxicity and neuronal death caused by ROS [12]. Following this rationale, we designed and synthesized memantine-FA conjugates following the two routes shown in Fig. 1.

It is well known that memantine's primary amine plays a crucial role in receptor binding [13]. Thus, to preserve this moiety, we functionalized the adamantane nucleus of memantine with a carboxylic function, which acted as the reactive point for FA conjugation, affording compounds 1–3. Further, in compounds 4–7, we explored the possibility to introduce FA appendages on the nitrogen atom of memantine, whose conversion to a secondary amine has previously emerged as a feasible strategy to gain memantine-based NMDAR antagonists [14,15]. Synthesized compounds were first tested against NMDAR. Based on their NMDAR blocking properties, compounds were selected to study their direct and indirect antioxidant efficacy, as well as the ability to modulate the amyloidogenic pathway.

## 2. Results and discussion

### 2.1. Chemistry

Memantine-FA hybrids 1–7 were prepared following two different synthetic routes (Schemes 1 and 2), depending on the way the two pharmacophores were connected. For the synthesis of compounds 1–3, the appropriate mono Boc-protected diamine (8–10) [16,17] was condensed with FA to give intermediates 11–13. Cleavage of the protecting group in acidic conditions led to compounds 14–16. Conjugation of 14–16 with 17 hydrochloride, which was obtained following a Ritter-type protocol as previously reported by Wanka et al. [18], afforded final compounds 1–3 (Scheme 1). To gain compounds 4–7, memantine hydrochloride (MEM) was

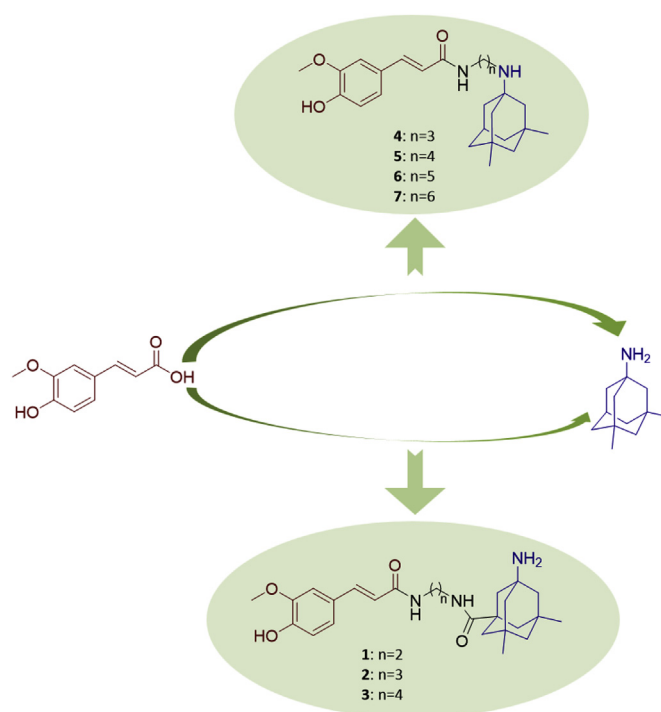


Fig. 1. Drug design of compounds 1–7.

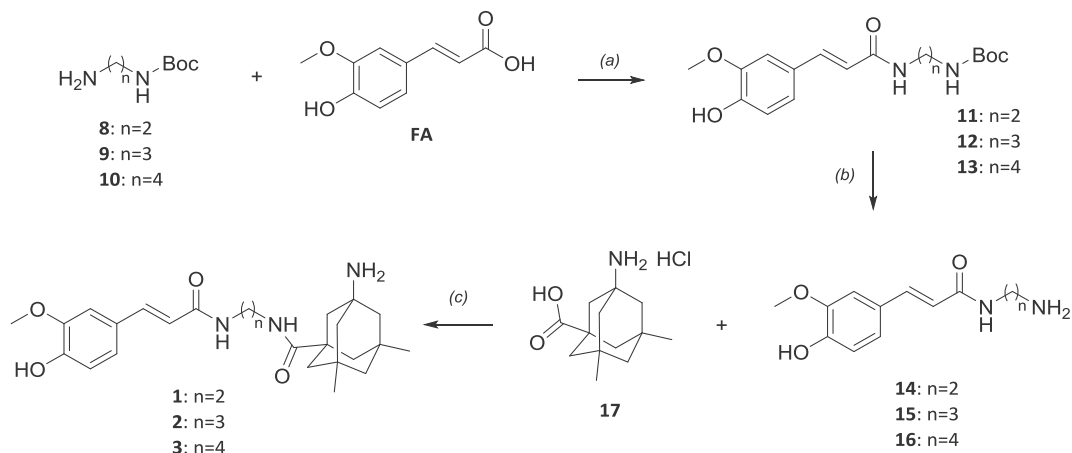
alkylated with the appropriate tosyl-activated alcohol (18–21) under basic conditions to give intermediates 22–25 which, after carbamate deprotection (26–29), were coupled with FA in the presence of EDC and HOBt (Scheme 2).

### 2.2. NMDAR blocking activity

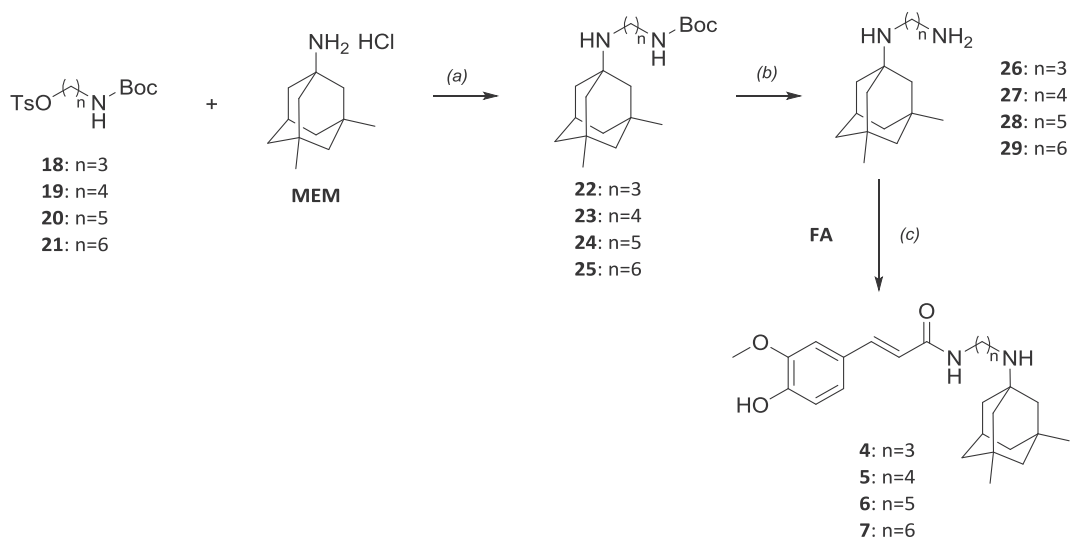
All the compounds were initially investigated to assess their effect at NMDAR. In particular, the antagonism of responses to NMDA and glycine were measured by voltage-clamp recordings on GluN1-1a/GluN2A NMDAR expressed in *Xenopus laevis* oocytes at  $-60$  mV, with compounds 1–7 applied in tenfold increments in the range  $0.01$ – $100$   $\mu$ M. Memantine was used as the reference compound. Compounds 1–3, which retain the primary amine function of memantine, demonstrated very low or no potency to block NMDAR (Fig. 2A). Conversely, employing memantine's amine for connecting FA appendages resulted in significant blocking of NMDA/glycine responses (Fig. 2A).

Compounds 5–7 presented a micromolar profile, with  $IC_{50}$  values ranging from  $6.9$  to  $23.9$   $\mu$ M, while the shorter compound 4 had an  $IC_{50}$  greater than  $100$   $\mu$ M, thus suggesting its inefficacy (Table 1). Blocking properties toward NMDAR were influenced by the chain length separating the pharmacophoric functions, with compound 7, carrying a hexamethylene spacer, emerging as the most efficacious. Compounds with considerable blocking properties (5–7) were assessed for voltage dependency. The compounds were diluted to their approximate  $IC_{50}$  concentrations and block of NMDA/glycine responses mediated by GluN1-1a/GluN2A was measured at four different holding potentials ( $-40$ ,  $-60$ ,  $-80$  and  $-100$  mV).

Compounds presented a voltage-dependent behavior, acting, like memantine, as open channel blockers of the receptor. Data were fitted with the Woodhull equation to determine their  $\delta$  values, and thus estimate the position of the binding site within the membrane electric field [19,20]. The results of this study showed that three of the new molecules yielded  $\delta$  values comparable to



**Scheme 1.** Reagents and conditions: (a) EDC, HOBT, DMF, Et<sub>3</sub>N, N<sub>2</sub>, 12 h, 0°C-rt; (b) HCl 4 M in dioxane, CH<sub>2</sub>Cl<sub>2</sub>, 90', 0°C-rt; (c) EDC, HOBT, DMF, N<sub>2</sub>, 36 h, 0°C-rt.



**Scheme 2.** Reagents and conditions: (a) K<sub>2</sub>CO<sub>3</sub>, KI, DMF, 140 °C, 1h, MW; (b) HCl 4 M in dioxane, CH<sub>2</sub>Cl<sub>2</sub>, 20', 0°C-rt; (c) EDC, HOBT, Et<sub>3</sub>N, DMF, N<sub>2</sub>, 12 h, 0°C-rt.

memantine. Compounds **5**, **6** and **7** had  $\delta$  values in the range 0.43–0.51, which are just slightly higher to that of memantine, 0.39. Based on their  $\delta$  values, we can suggest these compounds may have a binding site midway through the pore, maybe a little deeper but overlapping with that of memantine. This is consistent with binding adjacent to the Q/R/N-site that determines ion selectivity in ionotropic glutamate receptors.

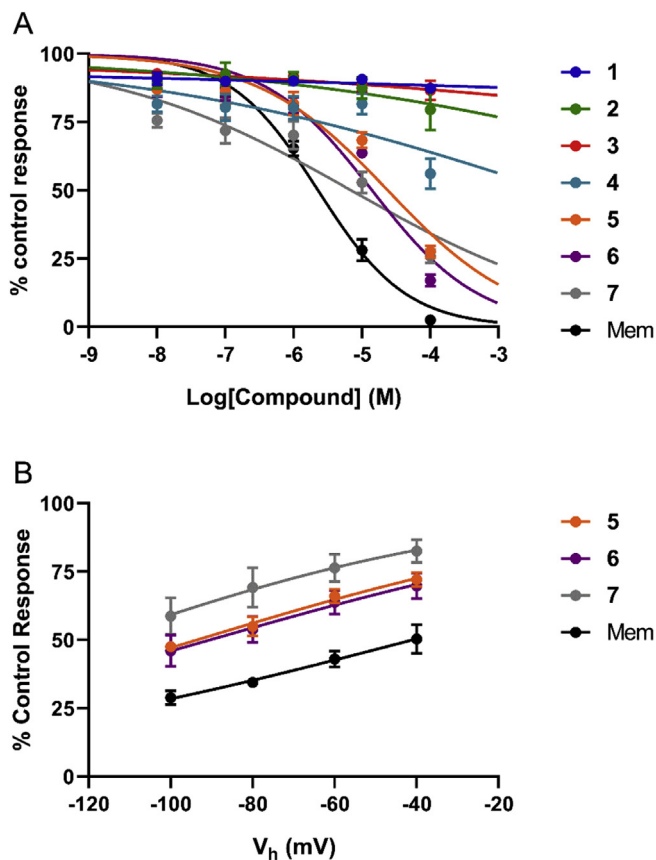
### 2.3. Cell toxicity assay

Compounds **5–7**, presenting appreciable NMDAR blocking properties, were selected for deepening their antioxidant profile in SH-SY5Y human neuroblastoma cells. To this aim, we assessed the cytotoxicity of compounds **5–7** to define the concentration range to be used in cellular experimental settings. The antioxidant FA was used for comparison. Cells were exposed to the compounds at concentrations ranging from 1 to 50  $\mu$ M for 24 h and cell viability was determined by MTT assay. As shown in Fig. 3, all the compounds were devoid of any toxicity at a concentration up to 20  $\mu$ M, while only the shorter derivative **5** retained, like FA, good tolerability up to 50  $\mu$ M. Lack of toxicity was verified also for compound **4**, carrying a three methylene spacer, at all the concentrations investigated (data not shown), confirming that the spacer length

significantly influenced compound tolerability in favor of shorter derivatives.

### 2.4. Protective effect toward H<sub>2</sub>O<sub>2</sub>-Induced damage

To determine the antioxidant efficacy of compounds **5–7**, we first studied their ROS scavenging activity when coincubated with 300  $\mu$ M H<sub>2</sub>O<sub>2</sub>, using FA for comparison. The scavenger effect was evaluated in SH-SY5Y cells by using the fluorescent probe dichlorodihydrofluorescein diacetate (DCFH-DA) as a marker for quantitative intracellular ROS formation. The DCFH-fluorescence intensity significantly increased in H<sub>2</sub>O<sub>2</sub>-treated cells (black line, Fig. 4A) with respect to untreated cells (dashed grey line, Fig. 4A). All compounds, at a concentration of 10  $\mu$ M, were able to markedly reduce H<sub>2</sub>O<sub>2</sub>-induced intracellular ROS formation, being, however, less effective than FA. To assess if indirect antioxidant effects could accompany radical scavenger properties, further experiments were performed pretreating SH-SY5Y cells with compounds **5–7** (10  $\mu$ M) for 24 h before adding 300  $\mu$ M H<sub>2</sub>O<sub>2</sub> (Fig. 4B). Again, compounds **5–7** produced a significant reduction in DCFH-fluorescent intensity, albeit an inversion in the trend of efficacy was observed. Indeed, with this experimental setting, compound **7** emerged as the most efficacious, reaching FA ability to counteract H<sub>2</sub>O<sub>2</sub>-induced



**Fig. 2.** A) Concentration-inhibition curves for compounds 1–7 in comparison to memantine (Mem). Data are mean % of control response to 100  $\mu$ M NMDA (+10  $\mu$ M glycine)  $\pm$  SEM (n = 5–7 separate oocytes). The curves are fits to Equation (1) and  $IC_{50}$  values are given in Table 1. B) Voltage dependence of inhibition by compounds 5–7 (30  $\mu$ M, 20  $\mu$ M and 10  $\mu$ M respectively) in comparison to memantine (Mem; 3  $\mu$ M). Data are plotted as mean % control response to 100  $\mu$ M NMDA (+10  $\mu$ M glycine)  $\pm$  SEM against the holding potential ( $V_h$ ) (n = 5–6 separate oocytes). The curves are fits to Equation (2) and  $\delta$  values are given in Table 1.

**Table 1**  
 $IC_{50}$  and  $\delta$  values for compounds 1–7 and memantine.

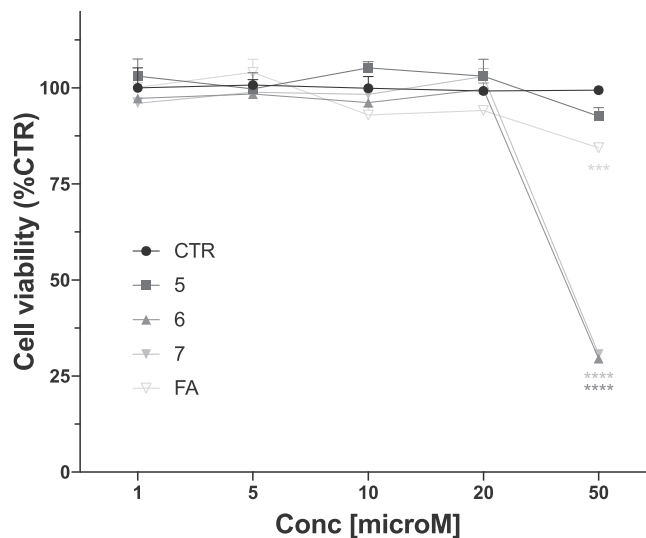
Compound	$IC_{50}$ [95% CI] $\mu$ M (n)	$\delta \pm SE$ (n)
1	$\gg 100$ (7)	nd
2	$\gg 100$ (6)	nd
3	$\gg 100$ (6)	nd
4	$>100$ (5)	nd
5	23.9 [13.0–49.4] (5)	$0.46 \pm 0.07$ (6)
6	14.1 [8.7–22.6] (5)	$0.43 \pm 0.12$ (6)
7	6.9 [3.0–19.2] (5)	$0.51 \pm 0.17$ (6)
memantine	2.3 [1.7–3.0] (6)	$0.39 \pm 0.08$ (5)

nd = not determined (because inhibition was too weak).

ROS formation. Based on these results, we could speculate that, at least for 7, antioxidant properties might derive from both direct and indirect effects.

### 2.5. Activation of Nrf2 pathway in SH-SY5Y cells

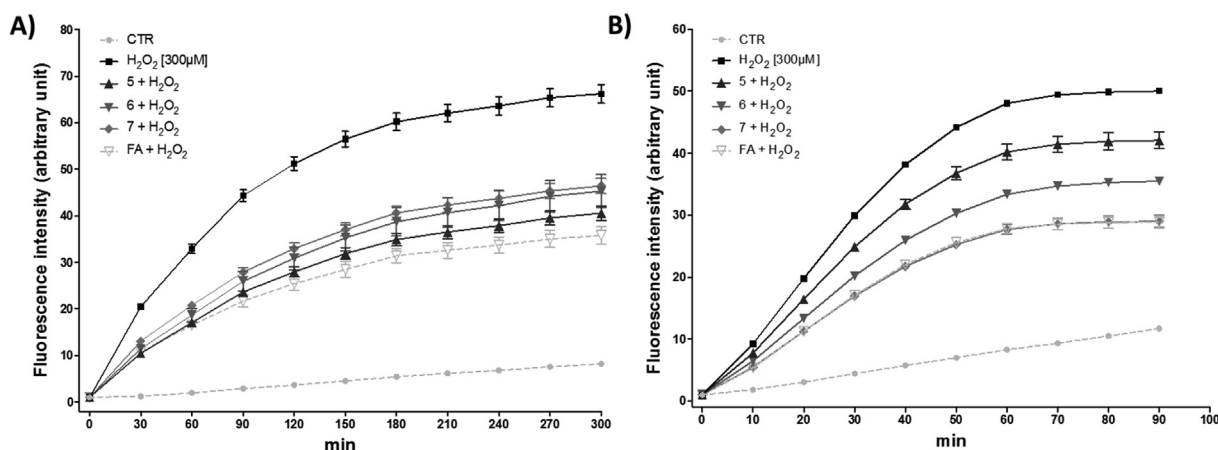
The nuclear factor (erythroid-derived 2)-like 2 (Nrf2) transcriptional pathway is a major player of inducible antioxidant defense [21]. Activation of the Nrf2 pathway, and the subsequent transcription of downstream cytoprotective genes, is triggered by the disruption of interaction and binding of Nrf2 with the cytosolic Nrf2 repressor Kelch-like ECH-associated protein 1 (Keap1) [22]. A



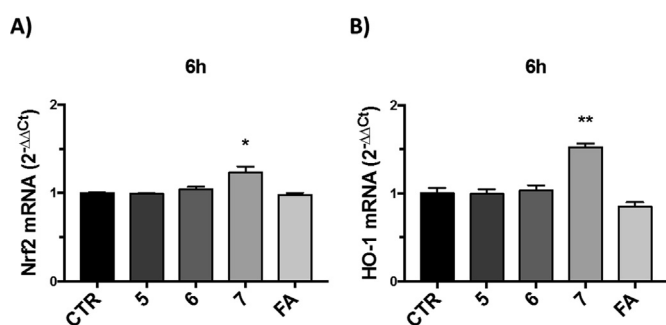
**Fig. 3.** Cellular toxicity of hybrid compounds (5–7) and Ferulic Acid (FA) on human neuroblastoma SH-SY5Y. Cells were treated with compounds for 24 h at different concentrations ranging from 1 to 50  $\mu$ M. Cell viability was assessed by MTT assay. Data are expressed as percentage of cell viability versus CTR; \*\*\* $p < 0.001$ , and \*\*\*\* $p < 0.0001$  versus CTR; Dunnett's multiple comparison test, n = 3.

variety of electrophiles from synthetic or natural sources is emerging for their ability to hamper this interaction by targeting key cysteine residues of Keap1, which act as sensors of oxidative insults [23]. In particular, the electrophilic motif recurring in FA and its derivatives, namely the  $\alpha,\beta$ -unsaturated carbonyl group, has already been shown to trigger the Nrf2-driven transcriptional process in a series of hydroxy-cinnamic derivatives for which trapping Keap1 through covalent adduct formation was proposed to be the initiating event [24,25]. Thus, we studied compounds 5–7 and FA in SH-SY5Y neuroblastoma cells to verify whether they may affect the Nrf2 pathway and indirect mechanisms could therefore contribute to their overall antioxidant profile. To this aim, we first assessed their ability to modulate the mRNA levels of Nrf2 by real-time PCR, using 10  $\mu$ M of each compound incubated for 6 h. Notably, only compound 7 determined a significant increase in Nrf2 mRNA expression (Fig. 5, panel A), while cells treated with FA or compounds 5 and 6 behaved like untreated cells. Coherently with these results, the same trend was observed when we investigated the ability of compounds to tune the mRNA levels of heme oxygenase-1 (HO-1), a prototypical Nrf2-target gene related to oxidative stress response. Indeed, mRNA levels of the inducible cytoprotective gene raised to about 150% of control following pre-treatment with 10  $\mu$ M 7, while no effect was elicited by FA or compounds 5 and 6 (Fig. 5, panel B).

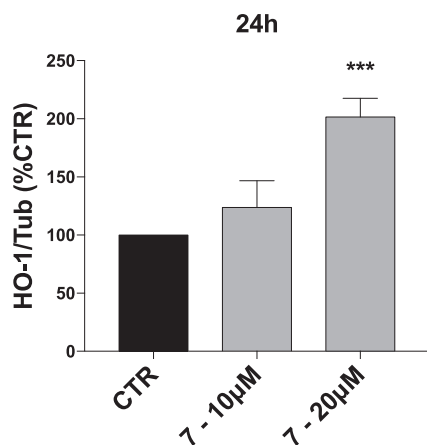
Then, we sought to verify whether the increase in HO-1 mRNA expression determined by compound 7 could effectively result in enhanced HO-1 protein levels. To this aim, HO-1 induction was analyzed by means of Western immunoblotting in the same cell line after treatment for 24 h with 7 at 10 or 20  $\mu$ M. Interestingly, compound 7 caused a dose-dependent increase of HO-1 expression, with cells treated with 20  $\mu$ M 7 almost doubling HO-1 protein levels of control (Fig. 6). These results confirm that compound 7 is a multimodal antioxidant, which combines radical scavenging properties to the ability of potentiating the Nrf2/HO-1 axis. Further, the lack of indirect antioxidant efficacy verified for compounds 5, 6 and FA, all carrying the  $\alpha,\beta$ -unsaturated carbonyl feature, reveal that an electrophilic moiety is not per se sufficient for activating redox sensor proteins, and shape complementarity may play a pivotal role in this respect. Particularly, we might speculate that



**Fig. 4.** Hybrid compounds reverse ROS formation induced by H<sub>2</sub>O<sub>2</sub>. (A) After the loading with DCFH-DA, SH-SY5Y cells were exposed to 10 μM compounds or FA in combination with 300 μM of H<sub>2</sub>O<sub>2</sub>. The fluorescence intensity for all compounds tested is significant at any time starting from 60 to 300 min with  $p < 0.0001$  versus H<sub>2</sub>O<sub>2</sub>. At 30 min the significance versus H<sub>2</sub>O<sub>2</sub> is  $p < 0.01$  for compound **7**,  $p < 0.001$  for compound **6** and  $p < 0.0001$  for compound **5** and FA. Dunnett's multiple comparison test. (B) SH-SY5Y cells were pre-treated with 10 μM of each compound for 24 h, loaded with DCFH-DA and then exposed to 300 μM H<sub>2</sub>O<sub>2</sub>. Fluorescence intensity for all compounds tested is significant at any time from 30 to 90 min with  $p < 0.0001$  versus H<sub>2</sub>O<sub>2</sub>. At time 10 min, the fluorescence intensity did not reach statistical significance for compound **5**, whereas for **6** the significance is  $p < 0.01$  and for **7** and FA is  $p < 0.0001$  vs H<sub>2</sub>O<sub>2</sub>. At 20 min, the significance is  $p < 0.001$  for compound **5** and  $p < 0.0001$  for compound **7** and FA vs H<sub>2</sub>O<sub>2</sub>. Dunnett's multiple comparison test.



**Fig. 5.** Hybrid compounds modulate Nrf2 and HO-1 mRNA levels. RNA was obtained from cellular extracts of SH-SY5Y cells treated for 6h with compounds **5–7** and FA at 10 μM and analyzed for Nrf2 (A) and HO-1 (B) mRNA expression by RT-PCR. GAPDH was used as housekeeping gene. Results are shown as mean ± SEM; \* $p < 0.05$ , \*\* $p < 0.01$  versus CTR; Dunnett's multiple comparison test,  $n = 3$ .



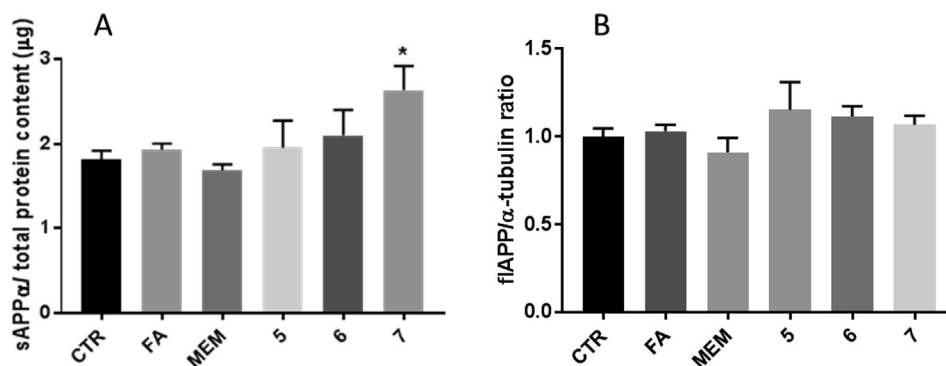
**Fig. 6.** Effect of compound **7** on HO-1 protein levels. Cellular extracts of SH-SY5Y cells treated for 24 h with compound **7** at 10 or 20 μM were analyzed for HO-1 protein levels by Western Blot. Anti-tubulin was used as protein loading control. Results are shown as ratio (% of CTR) ± SEM; \*\*\* $p < 0.001$  versus CTR; Dunnett's multiple comparison test,  $n = 3$ .

compounds **5–7**, varying in the linker length, and FA could differently orient their cysteine-reactive group toward nucleophilic traps of Keap1 affecting target recognition and, consequently, a compound's reactivity and specificity.

## 2.6. APP processing in H4-SW cells

In AD, a direct link exists between eNMDAR overactivation and increased neuronal A $\beta$  production [26]. NMDAR have been proposed to modulate  $\alpha$ -secretase activity, shifting APP metabolism towards a non-amyloidogenic pathway. Memantine has been shown to lower A $\beta$  synthesis in a number of studies [27,28]. Mechanisms potentially involved in memantine-driven A $\beta$  modulation are not completely clear, and both NMDAR-mediated and NMDAR-independent mechanisms have been proposed [29]. In this context, we sought to investigate whether the most promising compounds **5–7** could affect the APP processing favoring the production of the non-amyloidogenic soluble amyloid precursor protein  $\alpha$  (sAPP $\alpha$ ). Current research suggests that sAPP $\alpha$  plays a role in synaptic growth and plasticity, featuring neuroprotective and neurotrophic properties [30]. Thus, we studied the effect of the compounds on sAPP $\alpha$  levels in the human H4 cell line expressing the Swedish mutant form of APP (H4-SW), after 24 h treatment. Compounds **5–7** were tested at 10 μM concentration, which had no impact on cell viability, as confirmed by a dose-response curve where memantine, FA and compounds **5–7** had no toxic effect up to 20 μM concentration (data not shown). Memantine and FA alone were used for comparison. The Western blot analysis reported in Fig. 7 shows that compound **7**, but not compounds **5** and **6**, significantly increased sAPP $\alpha$  levels (Fig. 7, panel A).

Notably, compounds **5–7** were not able to affect full-length APP (fl-APP) expression levels, which was determined in the same cells to evaluate the effect of the compounds on total intracellular APP (Fig. 7, panel B). By enhancing sAPP $\alpha$  levels without affecting the levels of total intracellular APP, compound **7** seems to stimulate APP processing towards the  $\alpha$ -secretase (non-amyloidogenic) pathway, which should result in decreased A $\beta$  production. Noteworthy, at the same concentration, memantine and FA alone showed no effect on APP processing, strengthening the value of the design of a hybrid molecule. The lack of efficacy of memantine, whose potency as NMDAR antagonist is 3-fold higher than that of **7**, suggests that the



**Fig. 7.** Assessment of the effect of compounds 5–7, MEM and FA on amyloid precursor protein (APP) proteolytic processing in H4-SW cells. **(A) Determination of sAPP $\alpha$  levels.** Cells were treated with 10  $\mu$ M of each compound and after 24 h conditioned media were collected and sAPP $\alpha$  content assessed by Western blotting. The graph shows the densitometric quantification of the Western blotting bands, normalized to the total protein content of plated cells. \* =  $p < 0.05$ , one-way ANOVA and post-hoc test. **(B) Determination of full-length amyloid precursor protein (flAPP) expression.** Cells were treated with 10  $\mu$ M of each compound. After 24 h, H4-SW were lysed and flAPP expression assessed by Western blotting. The graph shows the densitometric quantification of the Western blotting bands, normalized to  $\alpha$ -tubulin as internal reference.

effect of **7** on APP processing we observed in this experimental setting seems to be not principally mediated by NMDAR. Interestingly, lengthening of the linker between the pharmacophoric functions up to six methylenes switched on the efficacy toward both APP processing and Nrf2 activation, pointing to **7** as the most promising molecule of the series.

### 3. Conclusions

NMDAR play a crucial role in the pathophysiology of AD. Excessive activation of NMDAR can compromise synapse function by triggering neurotoxic events, which involve A $\beta$  peptide and oxidative stress. By preferentially blocking extrasynaptic rather than synaptic currents, the anti-AD drug memantine limits neurotoxicity mediated by excessive NMDAR activity while relatively sparing physiological neurotransmission. This peculiar NMDAR profile prompted us to conjugate memantine with the bioactive payload FA, aiming to synergistically modulate the critical partnership occurring between oxidative damage, A $\beta$  burden, and hyperfunctioning NMDAR. For compounds **4–7**, which exploit memantine's nitrogen for FA connection, chain lengthening positively influenced NMDAR blocking properties. The longer derivative **7**, carrying a hexamethylene spacer between the pharmacophoric functions, presented a micromolar profile as NMDAR antagonist ( $IC_{50} = 6.9 \mu$ M), being only three times less effective than the parent compound memantine ( $IC_{50} = 2.3 \mu$ M). Further, compound **7** also shares with memantine the binding site midway through the pore and a voltage-dependent behavior, suggesting that conjugation with FA produced only a modest perturbation of memantine's NMDAR binding mode. Compounds with appreciable NMDAR blocking properties were studied in SH-SY5Y cells to assess their antioxidant properties. All compounds tested showed notable free radical scavenging effects. Conversely, only **7** was able to significantly potentiate the expression of Nrf2 and its downstream protective gene HO-1 at the concentration of 10  $\mu$ M, therefore emerging as a multimodal antioxidant. Notably, the lack of indirect antioxidant efficacy observed for **5** and **6**, varying in the linker length, and FA, suggests the importance of target recognition as a pre-requisite for electrophile reactivity, excluding an indiscriminate effect driven by the  $\alpha,\beta$ -unsaturated carbonyl group. At the same concentration (10  $\mu$ M), compound **7**, and not shorter derivatives **5** and **6**, significantly enhanced sAPP $\alpha$  levels in H4-SW cells, suggesting that it may stimulate APP processing in favor of the  $\alpha$ -secretase (non-amyloidogenic) pathway and consequently limit A $\beta$  formation. Thus, the most potent NMDAR antagonist **7** was

also able to activate inducible protective pathways which play a crucial role in contrasting the neurotoxic cascade driven by eNMDAR overactivation. The multimodal profile of compound **7** was well balanced, in the micromolar-range, and not accompanied by any cytotoxicity in both SH-SY5Y and H4-SW cells up to the concentration of 20  $\mu$ M. Based on these findings, compound **7** emerges as a promising pharmacologic tool for deepening our insight on the significance of NMDAR-mediated neurotoxic events involving ROS formation and A $\beta$  damage.

### 4. Experimental section

#### 4.1. Chemistry

Chemical reagents were purchased from Sigma-Aldrich, Fluka and Lancaster (Italy) and used without further purification. Chromatographic separations were performed on silica gel columns (Kieselgel 40, 0.040–0.063 mm, Merck) by chromatography. Reactions were followed by TLC on Merck (0.25 mm) glass-packed precoated silica gel plates (60 F254), then visualized with an UV lamp, bromocresol green or KMnO $_4$ . Melting points were measured in glass capillary tubes on a Büchi SMP-20 apparatus and are uncorrected. Microwave assisted synthesis was performed by using CEM Discover<sup>®</sup> SP apparatus (2.45 GHz, maximum power of 300 W). NMR spectra were recorded at 400 MHz for  $^1$ H and 100 MHz for  $^{13}$ C on Varian VXR 400 spectrometer. Chemical shifts ( $\delta$ ) are reported in parts per millions (ppm) relative to tetramethylsilane (TMS), and spin multiplicities are given as s (singlet), br s (broad singlet), d (doublet), t (triplet), q (quartet), or m (multiplet). Direct infusion ESI-MS mass spectra were recorded on a Waters ZQ 4000 apparatus. Final compounds **1–7** were >95% pure as determined by HPLC analyses. The analyses were performed under reversed-phase conditions on a Phenomenex Jupiter C18 (150  $\times$  4.6 mm I.D.) column, using a binary mixture of 0.1% TFA in H $_2$ O/acetonitrile (70/30, v/v for **5**; 65/35, v/v for **3, 4, 6, 7**; 80/20, v/v for **1, 2**) as the mobile phase, UV detection at  $\lambda = 302$  nm and a flow rate of 1 mL/min. Analyses were performed on a liquid chromatograph model PU-2089Plus UV equipped with a 20  $\mu$ L loop valve (Jasco Europe, Italy). Compounds were named relying on the naming algorithm developed by CambridgeSoft Corporation and used in Chem-BioDrawUltra 15.1.

#### 4.1.1. General procedure for the intermediates **11–13**

To an ice-cooled solution of ferulic acid (**FA**, 1 equiv) in dry DMF (3–4 mL) were added HOBT (1.3 equiv) and EDC (1.3 equiv). The

reaction mixture was stirred for 10 min, followed by addition of Et<sub>3</sub>N (1.3 equiv) and the appropriate mono-protected diamine (**8–10**) (1 equiv). Stirring was then continued at room temperature overnight, and the solvent evaporated under vacuum. The crude was purified by flash chromatography on silica gel using dichloromethane/methanol (9.5:0.5) as mobile phase.

#### 4.1.2. *tert*-butyl (*E*)-(2-(3-(4-hydroxy-3-methoxyphenyl)acrylamido)ethyl)carbamate (**11**)

Synthesized from **FA** (400 mg, 2.06 mmol) and **8** [17] (330 mg, 2.06 mmol) to afford **11** as waxy solid: 200 mg (30%); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.50 (d, *J* = 15.6 Hz, 1H), 6.99 (d, *J* = 8 Hz, 1H), 6.94 (s, 1H), 6.88 (d, *J* = 8.4 Hz, 1H), 6.60 (br s, 1H), 6.26 (d, *J* = 15.6 Hz, 1H), 5.33 (br s, 1H), 3.85 (s, 3H), 3.50–3.46 (m, 2H), 3.33–3.30 (m, 2H), 1.42 (s, 9H).

#### 4.1.3. *tert*-butyl (*E*)-(3-(3-(4-hydroxy-3-methoxyphenyl)acrylamido)propyl)carbamate (**12**)

Synthesized from **FA** (100 mg, 0.51 mmol) and **9** [16] (174 mg, 0.51 mmol) to afford **12** as waxy solid: 100 mg (56%); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.55 (d, *J* = 15.2 Hz, 1H), 7.05 (d, *J* = 8 Hz, 1H), 7.01 (s, 1H), 6.90 (d, *J* = 8 Hz, 1H), 6.50 (br s, 1H), 6.30 (d, *J* = 15.2 Hz, 1H), 4.96 (br s, 1H), 3.92 (s, 3H), 3.44–3.40 (m, 2H), 3.23–3.19 (m, 2H), 1.67–1.61 (m, 2H), 1.45 (s, 9H).

#### 4.1.4. *tert*-butyl (*E*)-(4-(3-(4-hydroxy-3-methoxyphenyl)acrylamido)butyl)carbamate (**13**)

Synthesized from **FA** (400 mg, 2.06 mmol) and **10** [31] (188 mg, 2.06 mmol) to afford **13** as waxy green solid: 230 mg (33%); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.52 (d, *J* = 15.2 Hz, 1H), 7.03 (d, *J* = 8.4 Hz, 1H), 6.98 (s, 1H), 6.88 (d, *J* = 8 Hz, 1H), 6.25 (d, *J* = 15.2 Hz, 1H), 6.01 (br s, 1H), 4.64 (br s, 1H), 3.89 (s, 3H), 3.39–3.38 (m, 2H), 3.17–3.13 (m, 2H), 1.60–1.55 (m, 4H), 1.43 (s, 9H).

#### 4.1.5. General procedure for the intermediates **14–16**

To an ice-cooled solution of the appropriate Boc-protected intermediate (**11–13**, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (2–3 mL) was added HCl 4 M in dioxane (2–3 mL) and the reaction mixture was stirred at 0 °C for 90 min. The solvent was evaporated and the crude purified by flash chromatography on silica gel using dichloromethane/methanol/aqueous ammonia 33% (8:2:0.2) affording desired intermediates as free bases.

#### 4.1.6. (*E*)-*N*-(2-aminoethyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**14**)

Synthesized from **11** (200 mg, 0.60 mmol) to afford **14** as pale yellow solid: 120 mg (86%); <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 8.34 (t, *J* = 5.4 Hz, 1H), 7.72 (d, *J* = 15.6 Hz, 1H), 7.53 (s, 1H), 7.41–7.38 (m, 1H), 7.20 (d, *J* = 7.6 Hz, 1H), 6.86 (d, *J* = 15.6 Hz, 1H), 4.21 (s, 3H), 3.59–3.55 (m, 2H), 3.04–3.00 (m, 2H), 2.92 (br s, 2H).

#### 4.1.7. (*E*)-*N*-(3-aminopropyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**15**)

Synthesized from **12** (100 mg, 0.30 mmol) to afford **15** as pale green solid: 71 mg (99%); <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD) δ 7.40 (d, *J* = 15.6 Hz, 1H), 7.10 (s, 1H), 7.00–6.98 (m, 1H), 6.77 (d, *J* = 8.4 Hz, 1H), 6.48 (d, *J* = 15.6 Hz, 1H), 3.84 (s, 3H), 3.38 (t, *J* = 6.4 Hz, 2H), 2.96 (t, *J* = 7.2 Hz, 2H), 1.95–1.88 (m, 2H).

#### 4.1.8. (*E*)-*N*-(4-aminobutyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**16**)

Synthesized from **13** (230 mg, 0.63 mmol) to afford **16** as pale green solid: 160 mg (96%); <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD) δ 7.39 (d, *J* = 15.6 Hz, 1H), 7.05 (s, 1H), 6.96 (d, *J* = 7.6 Hz, 1H), 6.73 (d, *J* = 8.4 Hz, 1H), 6.40 (d, *J* = 15.6 Hz, 1H), 3.82 (s, 3H), 3.31–3.23 (m,

2H), 2.80–2.78 (m, 2H), 1.59–1.58 (m, 4H).

#### 4.1.9. General procedure for compounds **1–3**

To an ice-cooled solution of the hydrochloride salt **17** [18] (1 equiv) in dry DMF (3 mL) were added HOBT (1.3 equiv) and EDC (1.3 equiv) under N<sub>2</sub> atmosphere. The reaction mixture was stirred for 10 min, followed by addition of the appropriate intermediates (**14–16**) (2 equiv). Stirring was continued at room temperature for 36–48 h, and then the solvent evaporated under vacuum. The crude was purified by column chromatography on silica gel using dichloromethane/methanol/aqueous ammonia 33% (8.5:1.5:0.15) as mobile phase.

#### 4.1.10. (*E*)-3-amino-*N*-(2-(3-(4-hydroxy-3-methoxyphenyl)acrylamido)ethyl)-5,7-dimethyladamantane-1-carboxamide (**1**)

Synthesized from **17** (64 mg, 0.24 mmol) and **14** (115 mg, 0.48 mmol) to afford **1** as green solid: 80 mg (74%); mp 122–124 °C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 7.99 (br s, 1H), 7.46 (br s, 1H), 7.32 (d, *J* = 15.6 Hz, 1H), 7.11 (s, 1H), 6.99–6.97 (m, 1H), 6.80 (d, *J* = 8.4 Hz, 1H), 6.40 (d, *J* = 15.6 Hz, 1H), 3.80 (s, 3H), 3.20–3.17 (m, 2H), 3.15–3.12 (m, 2H), 1.40 (s, 2H), 1.29–1.23 (m, 4H), 1.16–1.12 (m, 4H), 1.01 (s, 2H), 0.82 (s, 6H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ 176.77, 166.17, 148.84, 148.27, 139.54, 126.70, 121.97, 119.25, 116.10, 111.23, 55.96, 51.30, 49.90, 49.79, 49.03, 46.18, 44.55, 44.10, 38.90, 32.95, 30.19. MS [ESI+] *m/z* 442 [M+1]<sup>+</sup>.

#### 4.1.11. (*E*)-3-amino-*N*-(3-(3-(4-hydroxy-3-methoxyphenyl)acrylamido)propyl)-5,7-dimethyladamantane-1-carboxamide (**2**)

Synthesized from **17** (37 mg, 0.14 mmol) and **15** (71 mg, 0.28 mmol) to afford **2** as green solid: 31 mg (52%); mp 118–119 °C; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD) δ 7.43 (d, *J* = 16 Hz, 1H), 7.11 (s, 1H), 7.03–7.00 (m, 1H), 6.78 (d, *J* = 8 Hz, 1H), 6.41 (d, *J* = 16 Hz, 1H), 3.87 (s, 3H), 3.29 (t, *J* = 8 Hz, 2H), 3.22 (t, *J* = 8 Hz, 2H), 1.73–1.66 (m, 2H), 1.63 (s, 2H), 1.44 (s, 4H), 1.32–1.26 (m, 4H), 1.14 (s, 2H), 0.92 (s, 6H). <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>OD) δ 177.79, 168.00, 148.87, 148.00, 140.82, 126.55, 121.91, 117.08, 115.16, 110.09, 54.95, 50.48, 48.96, 48.41, 48.28, 44.09, 43.78, 43.43, 36.40, 36.37, 32.47, 28.98, 28.60. MS [ESI+] *m/z* 455 [M+1]<sup>+</sup>.

#### 4.1.12. (*E*)-3-amino-*N*-(4-(3-(4-hydroxy-3-methoxyphenyl)acrylamido)butyl)-5,7-dimethyladamantane-1-carboxamide (**3**)

Synthesized from **17** (89 mg, 0.30 mmol) and **16** (160 mg, 0.61 mmol) to afford **3** as green solid: 68 mg (48%); mp 116–117 °C; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD) δ 7.40 (d, *J* = 15.6 Hz, 1H), 7.05 (s, 1H), 6.98–6.95 (m, 1H), 6.73 (d, *J* = 8 Hz, 1H), 6.37 (d, *J* = 15.6 Hz, 1H), 3.82 (s, 3H), 3.27–3.24 (m, 2H), 3.18–3.15 (m, 2H), 1.54–1.49 (m, 6H), 1.38–1.36 (m, 4H), 1.24 (s, 4H), 1.07 (s, 2H), 0.86 (s, 6H). <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>OD) δ 177.94, 167.83, 149.83, 148.29, 140.76, 126.04, 122.02, 116.86, 115.46, 110.03, 54.92, 49.76, 49.12, 49.09, 48.47, 44.22, 44.09, 43.91, 38.77, 38.63, 32.46, 28.76, 26.53, 26.44. MS [ESI+] *m/z* 470 [M+1]<sup>+</sup>.

#### 4.1.13. General procedure for the intermediates **22–25**

A mixture of memantine hydrochloride (**MEM**, 1 equiv), K<sub>2</sub>CO<sub>3</sub> (2 equiv), KI (1 equiv) and the appropriate intermediate (**18–21**, 1 equiv) in dry DMF (2–5 mL) was placed in a microwave (140 °C, 250 Psi, 100 W) and left stirring for 1 h. The solvent was removed under reduced pressure and the crude purified by chromatography on silica gel using dichloromethane/methanol/aqueous ammonia 33% (9:1:0.2) as mobile phase.

#### 4.1.14. *tert*-butyl 3-((3,5-dimethyladamantan-1-yl)amino)propyl carbamate (**22**)

Synthesized from **18** [14] (0.4 g, 1.2 mmol) to afford **22** as a pale oil: 0.24 g (59%); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 5.34 (br s, 1H),

3.16–3.14 (m, 2H), 2.65 (t,  $J = 6.6$  Hz, 2H), 2.09–2.07 (m, 1H), 1.67–1.64 (m, 2H), 1.49 (s, 2H), 1.38 (s, 9H), 1.32–1.23 (m, 8H), 1.06–1.05 (m, 2H), 0.78 (s, 6H).

#### 4.1.15. *tert*-butyl (4-((3,5-dimethyladamantan-1-yl)amino)butyl) carbamate (**23**)

Synthesized from **19** [32] (0.4 g, 1.5 mmol) to afford **23** as a pale oil: 0.35 g (87%);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.32 (br s, 1H), 3.00–2.99 (m, 2H), 2.63 (t,  $J = 7.2$  Hz, 2H), 2.06–2.04 (m, 1H), 1.59–1.55 (m, 4H), 1.48–1.43 (m, 2H), 1.37–1.28 (m, 13H), 1.24–1.13 (m, 4H), 1.02–1.01 (m, 2H), 0.73 (s, 6H).

#### 4.1.16. *tert*-butyl (5-((3,5-dimethyladamantan-1-yl)amino)pentyl) carbamate (**24**)

Synthesized from **20** [33] (0.4 g, 1 mmol) to afford **24** as a pale oil: 0.22 g (55%);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.31 (br s, 1H), 3.11–3.09 (m, 2H), 2.59 (t,  $J = 7.4$  Hz, 2H), 2.15–2.13 (m, 1H), 1.54–1.42 (m, 15H), 1.40–1.25 (m, 10H), 1.16–1.11 (m, 2H), 0.85 (s, 6H).

#### 4.1.17. *tert*-butyl (6-((3,5-dimethyladamantan-1-yl)amino)hexyl) carbamate (**25**)

Synthesized from **21** [34] (0.6 g, 1.6 mmol) to afford **25** as a pale oil: 0.4 g (79%);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  4.61 (br s, 1H), 2.99–2.97 (m, 2H), 2.48 (t,  $J = 7.4$  Hz, 2H), 2.03–2.02 (m, 1H), 1.41–1.32 (m, 15H), 1.24–1.15 (m, 12H), 1.01–1.00 (m, 2H), 0.73 (s, 6H).

#### 4.1.18. General procedure for the intermediates **26–29**

To an ice-cooled appropriate Boc-protected intermediate (**22–25**, 1 equiv) was added HCl 4M in dioxane (4 mL) and the reaction mixture was stirred at 0 °C for 15–20 min under  $\text{N}_2$  atmosphere. The solvent was evaporated and the crude purified by flash chromatography on silica gel using dichloromethane/methanol/aqueous ammonia 33% (8:2:0.4) affording desired intermediates as free bases.

#### 4.1.19. $N^1$ -(3,5-dimethyladamantan-1-yl)propane-1,3-diamine (**26**)

Synthesized from **22** (0.24 g, 0.7 mmol) to afford **26** as a pale oil: 0.09 g (54%);  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta$  3.94 (br s, 3H), 2.74–2.73 (m, 2H), 2.59 (t,  $J = 6.8$  Hz, 2H), 2.06–2.04 (m, 1H), 1.57–1.50 (m, 2H), 1.42–1.41 (m, 2H), 1.26–1.18 (m, 8H), 1.09–1.00 (m, 2H), 0.78 (s, 6H).

#### 4.1.20. $N^1$ -(3,5-dimethyladamantan-1-yl)butane-1,4-diamine (**27**)

Synthesized from **23** (0.35 g, 1 mmol) to afford **27** as a pale oil: 0.18 g (72%);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  2.66–2.62 (m, 5H), 2.53 (t,  $J = 6.8$  Hz, 2H), 2.06–2.03 (m, 1H), 1.48–1.40 (m, 6H), 1.26–1.17 (m, 8H), 1.05–0.98 (m, 2H), 0.74 (s, 6H).

#### 4.1.21. $N^1$ -(3,5-dimethyladamantan-1-yl)pentane-1,5-diamine (**28**)

Synthesized from **24** (0.22 g, 0.6 mmol) to afford **28** as a pale oil: 0.11 g (70%);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  2.63 (t,  $J = 7.1$  Hz, 2H), 2.52 (t,  $J = 7.1$  Hz, 2H), 2.07–2.06 (m, 1H), 1.59 (br s, 3H), 1.42–1.38 (m, 6H), 1.31–1.17 (m, 10H), 1.05–1.04 (m, 2H), 0.78 (s, 6H).

#### 4.1.22. $N^1$ -(3,5-dimethyladamantan-1-yl)hexane-1,6-diamine (**29**)

Synthesized from **25** (0.4 g, 1.06 mmol) to afford **29** as a pale oil: 0.16 g (54%);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  2.63 (t,  $J = 7.2$  Hz, 2H), 2.52 (t,  $J = 7.2$  Hz, 2H), 2.38 (br s, 3H), 2.08–2.07 (m, 1H), 1.44–1.38 (m, 6H), 1.30–1.19 (m, 12H), 1.06–1.05 (m, 2H), 0.78 (s, 6H).

#### 4.1.23. General procedure for compounds **4–7**

To an ice-cooled solution of **FA** (1 equiv) in dry DMF (3–4 mL) was added HOBt (1.3 equiv) and EDC (1.3 equiv) under  $\text{N}_2$

atmosphere. The reaction mixture was stirred for 10 min at 0 °C, followed by addition of  $\text{Et}_3\text{N}$  (1.3 equiv) and the appropriate amine (**26–29**) (1 equiv). Stirring was then continued at room temperature overnight, then the solvent was evaporated under reduced pressure and the crude purified by chromatography on silica gel.

#### 4.1.24. (*E*)-*N*-(3-((3,5-dimethyladamantan-1-yl)amino)propyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**4**)

Synthesized from **26** (90 mg, 0.38 mmol), purified by chromatography on silica gel using petroleum ether/dichloromethane/methanol/aqueous ammonia 33% (2:6.5:1.5:0.16) as mobile phase to afford **4** as a yellow solid: 61.5 mg (39%); mp 213–215 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.66 (br s, 1H), 7.49 (d,  $J = 15.6$  Hz, 1H), 7.09 (s, 1H), 7.08–7.04 (m, 1H), 6.88 (d,  $J = 8$  Hz, 1H), 6.64 (d,  $J = 15.6$  Hz, 1H), 3.90 (s, 3H), 3.57–3.52 (m, 2H), 2.94–2.91 (m, 2H), 2.24 (m, 3H), 1.89–1.88 (m, 2H), 1.72–1.63 (m, 4H), 1.43–1.32 (m, 4H), 1.20 (s, 2H), 0.85 (s, 6H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  167.60, 147.53, 147.02, 140.33, 127.35, 122.15, 119.07, 114.87, 110.05, 59.04, 56.03, 49.88, 44.16, 41.92, 39.60, 38.59, 37.00, 32.57, 29.67, 26.77, 24.06. MS [ESI+]  $m/z$  413 [M+1]<sup>+</sup>.

#### 4.1.25. (*E*)-*N*-(4-((3,5-dimethyladamantan-1-yl)amino)butyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**5**)

Synthesized from **27** (72.5 mg, 0.28 mmol), purified by chromatography on silica gel using petroleum ether/dichloromethane/methanol/aqueous ammonia 33% (2:6.5:1.5:0.07) as mobile phase to afford **5** as a yellow solid: 39.6 mg (32%); mp 174–175 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.83 (br s, 1H), 7.41 (d,  $J = 15.6$  Hz, 1H), 7.04 (s, 1H), 6.95 (d,  $J = 8.2$  Hz, 1H), 6.86 (d,  $J = 8.2$  Hz, 1H), 6.70 (d,  $J = 15.6$  Hz, 1H), 3.81 (s, 3H), 3.33–3.32 (m, 2H), 2.82 (t,  $J = 7.4$  Hz, 2H), 2.17 (s, 1H), 1.89–1.88 (m, 4H), 1.71–1.62 (m, 4H), 1.37–1.23 (m, 6H), 1.14 (s, 2H), 0.84 (s, 6H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  167.60, 147.53, 147.02, 140.33, 127.35, 122.15, 119.07, 114.87, 110.05, 59.04, 56.03, 49.88, 44.16, 41.92, 39.60, 38.59, 37.00, 32.57, 29.67, 26.77, 24.06. MS [ESI+]  $m/z$  427 [M+1]<sup>+</sup>.

#### 4.1.26. (*E*)-*N*-(5-((3,5-dimethyladamantan-1-yl)amino)pentyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**6**)

Synthesized from **28** (54 mg, 0.2 mmol), purified by chromatography on silica gel using petroleum ether/dichloromethane/methanol/aqueous ammonia 33% (2:6.5:1.5:0.09) as mobile phase to afford **6** as a yellow solid: 27.7 mg (31%); mp 200–202 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.51 (d,  $J = 15.6$  Hz, 1H), 7.03 (d,  $J = 7.6$  Hz, 1H), 6.96 (s, 1H), 6.86 (d,  $J = 7.6$  Hz, 1H), 6.31 (d,  $J = 15.6$  Hz, 1H), 5.82 (br s, 1H), 3.86 (s, 3H), 3.37–3.33 (m, 2H), 2.61 (t,  $J = 7.2$  Hz, 2H), 2.13–2.11 (m, 1H), 1.58–1.53 (m, 4H), 1.36–1.24 (m, 12H), 1.10–1.09 (m, 2H), 0.82 (s, 6H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  166.21, 147.43, 146.81, 140.72, 127.33, 121.93, 118.40, 114.84, 109.72, 55.86, 50.83, 48.43, 42.89, 40.83, 40.37, 39.52, 32.37, 30.24, 30.19, 29.45, 24.78. MS [ESI+]  $m/z$  441 [M+1]<sup>+</sup>.

#### 4.1.27. (*E*)-*N*-(6-((3,5-dimethyladamantan-1-yl)amino)hexyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**7**)

Synthesized from **29** (0.1 g, 0.36 mmol), purified by chromatography on silica gel using petroleum ether/dichloromethane/methanol/aqueous ammonia 33% (2:6.5:1.5:0.1) as mobile phase to afford **7** as a yellow solid: 95.9 mg (59%); mp 203–204 °C;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.51 (d,  $J = 15.2$  Hz, 1H), 7.03 (d,  $J = 8.4$  Hz, 1H), 6.97 (s, 1H), 6.86 (d,  $J = 8.4$  Hz, 1H), 6.25 (d,  $J = 15.2$  Hz, 1H), 5.73 (br s, 1H), 3.88 (s, 3H), 3.36–3.31 (m, 2H), 2.57 (t,  $J = 7.4$  Hz, 2H), 2.12–2.09 (m, 1H), 1.54–1.45 (m, 6H), 1.33–1.23 (12H), 1.13–1.05 (m, 2H), 0.82 (s, 6H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  166.20, 147.57, 146.90, 140.74, 127.26, 121.96, 118.35, 114.90, 109.75, 55.85, 50.85, 48.42, 42.90, 40.79, 40.45, 39.54, 32.37, 30.24, 30.20, 29.50, 27.03, 26.73. MS [ESI+]  $m/z$  455 [M+1]<sup>+</sup>.



#### 4.2. Electrophysiological assays

Inhibition of NMDARs by compounds **1–7** and memantine was assessed by the expression of GluN1-1a and GluN2A subunits in *Xenopus* oocytes followed by voltage clamp recording. Oocytes were obtained from the European *Xenopus* Resource Centre (University of Portsmouth, UK) directly following their removal from mature female *Xenopus laevis* according to UK Home Office guidelines. Sections of the ovary were cut into smaller pieces and treated with 1 mg/mL collagenase type 1A (Sigma-Aldrich) in Ca<sup>2+</sup>-free modified Barth's media containing 96 mM NaCl, 2 mM KCl, 1 mM MgCl<sub>2</sub>, 5 mM HEPES, 2.5 mM pyruvic acid, 0.5 mM theophylline, 0.05 mg/mL gentamicin, pH 7.5, with shaking at 18 °C for 40–60 min, in order to separate them into individual defolliculated oocytes. The oocytes were then rinsed in Ca<sup>2+</sup>-free modified Barth's media multiple times until the solution was clear and stored in modified Barth's media (as per Ca<sup>2+</sup>-free but including 1.8 mM CaCl<sub>2</sub>). Oocytes were injected with 50 nL of cRNA encoding both the GluN1-1a and GluN2A subunits (1:1 by weight ratio; total 250 ng/μL). The cRNA was synthesized from linearized plasmid DNA (pRK7) containing the GluN-encoding genes using an mMessage mMachine kit (Invitrogen). Following injection, oocytes were kept in modified Barth's media at 18 °C for 3–4 days before electrophysiological testing. Voltage-clamp recording was conducted using an Axoclamp-2A voltage-clamp amplifier (Axon Instruments, USA). Microelectrodes were pulled from borosilicate glass capillaries (TW150F-4, World Precision Instruments) using a Sutter P-97 programmable micropipette puller to have a resistance of 0.5–2 MΩ when filled with 3 M KCl. Oocytes were placed in a perfusion chamber and constantly perfused (~5 mL/min) with solution containing 96 mM NaCl, 2 mM KCl, 1.8 mM CaCl<sub>2</sub>, 10 mM HEPES, pH 7.5, and voltage-clamped at holding potentials (V<sub>h</sub>) between –40 and –100 mV. NMDAR currents were initiated by application of 100 μM NMDA + 10 μM glycine. Once the current had reached a steady state (~30 s) the test compounds were introduced at concentrations ranging from 0.01 to 100 μM until a new plateau (inhibited current) was achieved. All agonists and test compounds were applied using an Automate Valvelink 8 perfusion system. Analogue output from the amplifier was digitized by a CED 1401 plus A/D converter at 100 Hz and recorded on a windows PC using WinEDR software (Dr John Dempster, University of Strathclyde, UK). NMDA/glycine-evoked current in the presence of test compound was normalized to that in its absence just before test compound addition (% control response) and plotted against concentration. Concentration-inhibition data were fit by:

$$\% \text{ control response} = \frac{100}{1 + 10^{((\log_{10} IC_{50} - X) \times \text{HillSlope})}} \quad (1)$$

to obtain IC<sub>50</sub> values, where X = Log<sub>10</sub>[compound]; using Graphpad Prism 7. All points were means of at least 5 separate oocytes. For voltage dependence studies the test compounds were applied at a single (~IC<sub>50</sub>) concentration but at four V<sub>h</sub>s in the range –40 to –100 mV. Data were normalized as above and fit by:

$$\% \text{ control response} = \frac{100}{\left(1 + [B]/K_D\right)^{z\delta} \left(\frac{FE}{RT}\right)} \quad (2)$$

to obtain δ values (fraction of the membrane electric field crossed by the blocking compound), where [B] is the concentration of the blocker, K<sub>D</sub> is the dissociation constant at 0 mV, z is the charge valence of the blocker, F is Faraday's constant, E is the membrane potential, R is the universal gas constant and T is

absolute temperature; using Graphpad Prism 7. All points were means of at least 5 separate oocytes.

#### 4.3. Reagents for cellular experiments (SH-SY5Y cells)

All hybrid compounds were solubilized in DMSO (at stock concentrations) and frozen (–20 °C) in aliquots that were diluted immediately prior to use. For each experimental setting, one stock aliquot was thawed out and diluted to minimize compound damage due to repeated freeze and thaw cycles. The final concentration of DMSO in culture medium was <0.1%. Ferulic Acid was purchased from Sigma Aldrich (Merck KGaA, Darmstadt, Germany). Rabbit polyclonal anti-human HO-1 (NBP1-31341) antibody was purchased from Novus (Biotechnie, Minneapolis USA). Mouse monoclonal anti-β-tubulin (T0198) was purchased from by Sigma Aldrich (Merck KGaA, Darmstadt, Germany).

#### 4.4. Cell cultures

All culture media, supplements and Foetal Bovine Serum (FBS) were purchased from Sigma Aldrich (Merck KGaA, Darmstadt, Germany). Human neuroblastoma SH-SY5Y cells from the European Collection of Cell Cultures (ECACC No. 94030304) were cultured in a medium with equal amounts of Eagle's minimum essential medium and Nutrient Mixture Ham's F-12, supplemented with 10% heat-inactivated FBS, 2 mM glutamine, 0.1 mg/mL streptomycin, 100 IU·mL penicillin and non-essential aminoacids at 37 °C in 5% CO<sub>2</sub> and 95% air atmosphere. H4-SW cells were cultured in D-MEM medium (Invitrogen, Carlsbad, CA) supplemented with 10% fetal bovine serum (FBS), 100 U/mL penicillin, 100 μg/mL streptomycin and 2 mM L-glutamine. Hygromycin B and Blastidicin S were used as selection antibiotics for SW mutation maintenance.

#### 4.5. Cell viability

The mitochondrial dehydrogenase activity that reduces 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-tetrazolium bromide (MTT, Sigma Aldrich, Merck KGaA, Darmstadt, Germany) was used to determine cellular viability, in a quantitative colorimetric assay. At day 0, SH-SY5Y cells were plated at a density of 2.5 × 10<sup>4</sup> viable cells per well in 96-well plates. After treatment, according to the experimental setting, cells were exposed to an MTT solution in complete medium (1 mg/mL). Following 4 h incubation with MTT and treatment with sodium dodecyl sulfate (SDS) for 24 h, cell viability reduction was quantified by using a Synergy HT multi-detection microplate reader (Bio-Tek).

#### 4.6. Measurement of intracellular ROS

DCFH-DA (Sigma Aldrich, Merck KGaA, Darmstadt, Germany) was used to estimate intracellular ROS following two different experimental setting described in each figure legend. In each setting, cells were loaded with 25 μM DCFH-DA for 45 min. After centrifugation DCFH-DA was removed, and the results were visualized using Synergy HT multidetection microplate reader (BioTek) with excitation and emission wavelengths of 485 and 530 nm, respectively.

#### 4.7. Real-time PCR (RT-PCR)

For RNA extraction, 2 × 10<sup>6</sup> cells were used. Total RNA was extracted using a Direct-zol™ RNA MiniPrep (Zymo Research, Irvine, USA) following the manufacturer's instructions. QuantiTect reversion transcription kit and QuantiTect Sybr Green PCR kit (Qiagen, Valencia, CA, USA) were used for cDNA synthesis and gene

expression analysis, following the manufacturer's specifications. Nrf2, HO-1, and GAPDH primers were provided by Qiagen. GAPDH was used as an endogenous reference.

#### 4.8. Immunodetection of HO-1, *fl*APP and *sAPP* $\alpha$

The expression of HO-1 in whole cell lysates was assessed by Western blot analysis. After treatment, cell monolayers were washed twice with ice-cold PBS, lysed on the culture dish by the addition of ice-cold homogenization buffer (50 mM Tris-HCl pH 7.4, 150 mM NaCl, 5 mM EDTA, 0.5% Triton X-100 and protease inhibitor mix) and an aliquot was used for protein quantification, whereas the remainder was prepared for Western blot by mixing the cell lysate with 2X sample buffer (125 mM Tris-HCl pH 6.8, 4% SDS, 20% glycerol, 6%  $\beta$ -mercaptoethanol, 0.1% bromophenol blue) and then denaturing at 95 °C for 5 min. Equivalent amounts of extracted proteins were loaded into a SDS-PAGE gel, electrophoresed under reducing conditions, transferred to a PVDF membrane (Sigma Aldrich, Merck KGaA, Darmstadt, Germany) and then blocked for 1 h with 5% w/v BSA in Tris-buffered saline containing 0.1% Tween 20 (TBS-T). The proteins were visualized using primary antibodies for HO-1, full length (*fl*) APP or soluble APP alpha (*sAPP* $\alpha$ ) and  $\alpha$ - or  $\beta$ -tubulin (1:1000) followed by secondary horseradish peroxidase conjugated antibody (1:5000) diluted in 5% w/v BSA in TBS-T. Tubulins were performed as a control for gel loading. Signal development was carried out using an enhanced chemiluminescent method (Pierce, Rockford, IL, USA).

#### 4.9. Densitometry and statistics

All experiments, unless specified, were performed at least three times with representative results being shown. Data are expressed as mean  $\pm$  SEM. The relative densities of the acquired images of Western blotting bands were analyzed with ImageJ software. Statistical analyses were performed using Prism software version 7.0 (GraphPad Software, La Jolla, CA, USA). Statistical differences were determined by analysis of variance (ANOVA) followed, when significant, by an appropriate post hoc test as indicated in figure legends. A *p* value of <0.05 was considered statistically significant.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.ejmech.2019.07.011>.

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